

REARRANGEMENT OF STEROID-14-ENE

Hiroko Izawa, Yoshimi Katada, Yukiko Sakamoto, and Yoshihiro Sato  
Kyoritsu College of Pharmacy, Minato-ku, Tokyo, Japan.

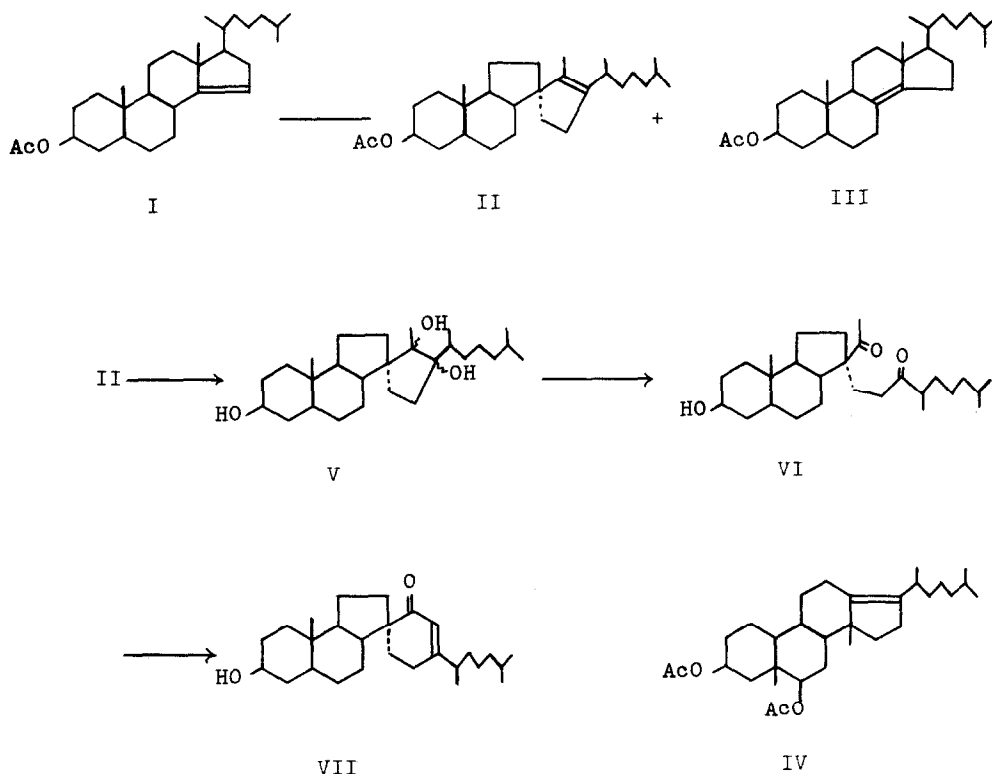
(Received in Japan 11 June 1969; received in UK for publication 20 June 1969)

Action of acids on some isolated double bonds such as 4-ene and 5-ene of steroids causes skeletal rearrangements (1). But, rearrangement of steroid-14-ene was not known. We wish to report a skeletal rearrangement reaction of steroid-14-ene.

Treatment of 3 $\beta$ -acetoxycholest-14-ene (I) with an equimolar boron trifluoride etherate in benzene and acetic anhydride for 170 hr., followed by chromatography on silica gel afforded an oily product (II) and a crystalline product (III), mp. 77-78°, in 23 and 6 % yield, respectively. Product (III) was proved to be 3 $\beta$ -acetoxycholest-8(14)-ene by direct comparison with an authentic specimen (2).

The oily product (II) was an isomer of I (C<sub>29</sub>H<sub>48</sub>O<sub>2</sub>, M<sup>+</sup> 428). II had an absorption maximum at 204 m $\mu$  ( $\epsilon$ =6900) in UV spectrum, which indicated the presence of a double bond. The nmr spectrum had methyl signals at  $\tau$  8.54 (t, J=0.8), 9.07 (d, J=7), 9.14(s), 9.20 (s) and 9.22 (s). The signal at  $\tau$  8.54 was assigned to a methyl group attaching to a double bond, which was found to be tetra-substituted from absence of a signal due to an olefinic proton. Coupling between the methyl signal at  $\tau$  9.07 and a signal at  $\tau$  7.54 (m, 1H) was confirmed by spin-spin decoupling experiment. Its mass spectrum showed a strong fragment (m/e 315, M<sup>+</sup>-C<sub>8</sub>H<sub>17</sub>) due to the fission of its side chain. This is not the case of steroids which have the side chains not attached to double bonds (3). A backbone rearrangement product (IV) possessing its side chain attached to 13(17)-ene was reported to show strong fragmentation of the side chain (4). Therefore, the side chain of II was considered to be attached to a double bond.

The product (II) was oxidized with osmium tetroxide in tetrahydrofuran, followed by reduction with lithium aluminum hydride to yield a mixture of isomeric



cis-triols (V). The mixture (V) was oxidized with lead tetraacetate to give diketone (VI), mp. 52.5-54°. The nmr spectrum of VI had a signal at  $\tau$  7.90 due to an acetyl group newly formed and indicated disappearance of the olefinic methyl signal. The IR spectrum showed carbonyl bands at 1712 and 1694  $\text{cm}^{-1}$ . When diketone (VI) was treated with sodium methoxide in methanol, an oily product (VII) (3,5-dinitrobenzoate of VII, mp. 168-170°) was obtained. The UV and IR spectra showed an absorption maximum at 240  $\mu$  ( $\epsilon=12800$ ) and a carbonyl band at 1650  $\text{cm}^{-1}$ , respectively. The nmr spectrum of the 3,5-dinitrobenzoate indicated absence of acetyl group and formation of one olefinic proton ( $\tau$  4.18, s). These data were consistent with the structure of  $\alpha,\beta$ -unsaturated ketone for VII, which was produced by aldol condensation of diketone (VI). Therefore, the structure having C/D ring spiran was determined for II. Since II was formed by migration of  $\text{C}_{12-13}$  carbon bond to  $\text{C}_{14}$ , the stereochemistry of the spiran ring was inherent.

Generally, reagents attack steroid-14-ene from rear side of the molecule. If it is true in this case, C<sub>14</sub> carbonium ion would be the intermediate, which would be converted to II by C<sub>12-13</sub> carbon bond migration and to III by elimination of C<sub>8</sub> proton.

Similar C<sub>12-13</sub> carbon bond migration to C<sub>14</sub> had been deduced for the products of elimination of 14 $\beta$ -hydroxy-15-oxo- and 14 $\beta$ -hydroxy-15 $\alpha$ -acetoxy-sterane acid derivatives with thionyl chloride (5).

#### Acknowledgment

We would like to express our gratitude to Professor Emeritus K. Tsuda of University of Tokyo, President of Kyoritsu College of Pharmacy, for his encouragement throughout this work. We thank to Professor S. Okuda of University of Tokyo and Dr. N. Ikekawa of Tokyo Institute of Technology for the measurements of nmr and mass spectra. We are also grateful to Sankyo Co. Ltd. for spin-spin decoupling experiment.

#### References

1. J. W. Blunt, M. P. Hartshorn and D. N. Kirk, Tetrahedron, **25**, 149 (1969).
2. O. Wiese, Chem. Ber., **69**, 2702 (1936).
3. L. Tökés, G. Jones and C. Djerassi, J. Am. Chem. Soc., **90**, 5465 (1968).
4. G. Snatske and H. Fehlhauer, Liebigs Ann. Chem., **676**, 188 (1964).  
J. W. Blunt, M. P. Hartshorn and D. N. Kirk, Chem. Comm., 160 (1966).
5. A. Lardon and T. Reichstein, Helv. Chim. Acta, **45**, 943 (1962).